Basic Analytical Techniques

CREDIT DISTRIBUTION, ELIGIBILITY AND PRE-REQUISITES OF THE COURSE

Course	Credits	Credit distribution of the course			Eligibility	Pre-requisite
title &		Lecture	Tutorial	Practical/	criteria	of the course
Code				Practice		(if any)
Basic	2	0	0	2	XII th Pass	NA
Analytical					with Science	
Techniques						

Learning Objectives

- To make students aware of the importance and the concepts of chemical analysis of water and soil samples collected from different sources
- To make them learn few techniques like chromatography, analytical techniques and instrumentation techniques, for example: spectrophotometry and flame photometry.

Learning Outcomes

By the end of the course, the students will be able to:

- Handle analytical data
- Determine the pH and conductance of soil samples, which can be useful in agriculture sector
- Do quantitative analysis of metal ions in water samples
- Separate ions using chromatographic techniques
- Estimate macronutrients using Flame photometry.

SYLLABUS

Practicals:

- 1. Determination of pH of soil samples collected from college nursery, sports ground and the soil collected from Yamuna River Bank.
- 2. Determination of conductance of soil samples collected from college nursery and sports ground.
- 3. Determination of pH of different types of aerated drinks and fruit juices.
- 4. Estimation of Calcium and Magnesium ions as Calcium carbonate (total hardness) by complexometric titration.
- 5. Determination of pH, acidity, and alkalinity of water samples collected from different water body/supply sources like Yamuna water, MCD supply water, Groundwater, water samples collected from water sewage treatment plants (Delhi /NCR).
- 6. Determination of dissolved oxygen (DO) of a water sample collected from different sources (at least two sources).
- 7. Determination of BOD of water sample collected from different water sources.
- 8. Paper chromatographic separation (*ascending and circular both*) of the mixture of metal ion (Ni²⁺ and Co²⁺) and (Cu²⁺ and Cd²⁺).
- 9. To study the use of phenolphthalein in trap cases.
- 10. Estimation of macro-nutrients: Potassium, calcium and magnesium in soil samples by flame photometry.
- 11. Spectrophotometric determination of Iron in vitamin / dietary tablets / different solutions of iron.
- 12. Spectrophotometric identification and determination of caffeine and benzoic acid in soft drink.
- 13. Spectrophotometric determination of cadmium and chromium in the given water sample.
- 14. Determination of ion exchange capacity of anion / cation exchange resin (using batch procedure if use of column is not feasible).
- 15. Visit STP plants and different chemical industries.

References:

- 1. Svehla, G. (1996), Vogel's Qualitative Inorganic Analysis, Prentice Hall.
- 2. Mendham, J.; Denney, R.C.; Barnes, J.D.; Thomas, M.J.K. (2007), Vogel's Quantitative Chemical Analysis, 6th Edition, Prentice Hall.
- 3. De, A. K. (2021), Environmental Chemistry, 10th edition. New Age International Pvt. Ltd. Note: Learners are advised to use the latest edition of readings.

(15 WEEKS)